

Available online at www.sciencedirect.com



Tetrahedron Letters 45 (2004) 3517-3520

Tetrahedron Letters

## Highly regio- and stereoselective PdCl<sub>2</sub>(MeCN)<sub>2</sub>-catalyzed cross coupling of 1,2-allenylic sulfoxides with allyl bromide

Shengming Ma,\* Qi Wei and Hongjun Ren

State Key Laboratory of Organometallic Chemistry, Shanghai Institute of Organic Chemistry, Chinese Academy of Sciences, 354 Fenglin Lu, Shanghai 200032, PR China

Received 16 January 2004; revised 19 February 2004; accepted 20 February 2004

Abstract—2-Allyl-1(E),3(E)-dienyl sulfoxides were prepared highly stereoselectively via the PdCl<sub>2</sub>(MeCN)<sub>2</sub>-catalyzed coupling reaction of 1,2-allenylic sulfoxides and allyl bromide. A rationale was proposed for this transformation. © 2004 Elsevier Ltd. All rights reserved.

It was reported that the reaction of an allene with  $PdCl_2(PhCN)_2$  usually afforded dinuclear monomeric 2-chloro- $\pi$ -allylic Pd **A**, dimeric  $\pi$ -allylic Pd complex **C**, or even trimeric  $\pi$ -allylic Pd complex **F** (Scheme 1).<sup>1-3</sup> It is quite interesting to note that similar to carbopalla-



Scheme 1.

7510; e-mail: masm@mail.sioc.ac.cn

dation of allenes<sup>4</sup> the initial halopalladation would also favor the formation of a  $\pi$ -allylic Pd complex of type **D** via the attack of the halogen atom at the center carbon atom. An exception was observed when PdCl<sub>2</sub>(PhCN)<sub>2</sub> was added as a solid to propadiene, in which the chlorine atom attacked the terminal carbon atom of propadiene forming a vinylic palladium species **B**.<sup>1a</sup> Another example is the catalytic carbonylation of allenes with PdCl<sub>2</sub> and CuCl<sub>2</sub> in MeOH, which afforded 2ethoxymethylacrylates via the intermediacy of **B** (Y = OMe) (Scheme 1).<sup>5</sup>

In principle, these two types of Pd intermediates, for example, **B** or **D**, may undergo further transformation with a variety of other reagents to afford useful synthetic intermediates providing that the regio- and stereoselectivity can be controlled. To the best of our knowledge, besides the carbonylation,<sup>5,6</sup> no report has been disclosed on the catalytic halopalladation-initiated coupling of an allene with a second partner. In this communication, we wish to disclose our recent observation on the Pd-catalyzed highly regio- and stereoselective coupling reaction of allyl bromide and 1,2-allenylic sulfoxides forming 2-allyl-1(*E*),3(*E*)-dienyl sulfoxides, in which the chlorine atom may attack the 3-position of 1,2-allenyl sulfoxides forming a vinylic Pd intermediate highly stereoselectively.

Recently during the course of systematic study of allene chemistry,<sup>7</sup> we are interested in the chemistry of 1,2-allenyl sulfoxides.<sup>8–10</sup> We observed that the stereoselectivity in the halohydroxylation of 1,2-allenyl sulfoxides was controlled by the participation of the sulfinyl group.<sup>9,10</sup> Based on these results we were interested to

*Keywords*: Palladium; 1,2-Allenyl sulfoxides; Allyl bromide; Coupling. \* Corresponding author. Tel.: +86-21-6416-3300; fax: +86-21-6416-

<sup>0040-4039/\$ -</sup> see front matter @ 2004 Elsevier Ltd. All rights reserved. doi:10.1016/j.tetlet.2004.02.144

see the effect of the sulfinyl group on the regio- and stereoselectivity of halometalation of 1,2-allenyl sulfoxides. We initiated this study by using 3-methyl-1,2butadienyl phenyl sulfoxide 1a with allyl bromide. Some typical results are summarized in Table 1. With RuCl<sub>3</sub> in the presence of LiOAc, the reaction afforded 3-methyl-2-allyl-1(E),3(E)-butadienyl sulfoxide **3a** in 16% yield (Table 1, entry 1). The stereochemistry was established by the NOESY spectra of 3a. It is interesting that the reaction with  $5 \mod \%$  of PdCl<sub>2</sub> afforded the product **3a** in 69% yield (Table 1, entry 2). The result with 5 mol% of PdCl<sub>2</sub>(MeCN)<sub>2</sub> and anhydrous LiOAc is slightly better (compare Table 1, entries 2-4). However, the presence of LiOAc is critical since the reaction in absence of LiOAc or in the presence of NaHCO3 afforded 3a only in very low yields (Table 1, entries 5 and 6). Furthermore it should be noted that no 3a was formed under the catalysis of a Pd(0) complex (Table 1, entries 7 and 8).

With the standard reaction conditions, some typical examples are shown in Table 2 with the following points noteworthy:

- (1) the stereoselectivity of 1,2-carbon-carbon double bond in product 3 ranges from 92:8 to 97:3 (E/Z);
- (2) the stereoselectivity of 3,4-carbon-carbon double bond in **3** is 100% E.

In the reaction of 1a with allyl bromide, besides the formation of 3a in 72% yield, 4% of 2-(3'-bromo-1'propenyl)-3-methylbuta-1,3-dienyl phenyl sulfoxide 4a was also formed, indicating that the reaction may involve a  $sp^2$  carbon-palladium intermediate **5a** (Scheme 2).

A rationale shown in Scheme 3 was proposed for the PdCl<sub>2</sub>(MeCN)<sub>2</sub>-catalyzed reaction of 1,2-allenyl sulfoxides with allyl bromide. In the first step the divalent palladium would undergo highly regioselective halopalladation reaction with the allene moiety of 1,2allenvlic sulfoxides to form the vinvlic intermediate 7. A LiOAc-mediated elimination reaction of Z-7 may lead to the formation of intermediate Z-5, which may undergo sequential carbopalladation and β-debromopalladation with allyl bromide to afford the final product 3.<sup>11</sup> The minor product 4a would be formed by the



Entry	Catalyst	Base	Solvent (2/THF)	Yield of <b>3a</b> (%)	Compound 1a recovered (%)
1	RuCl <sub>3</sub>	LiOAc	1:1	16	_
2	PdCl <sub>2</sub>	LiOAc	1:1	69	_
3	Pd(CH <sub>3</sub> CN) <sub>2</sub> Cl <sub>2</sub>	LiOAc·2H <sub>2</sub> O	1:1	70	_
4	Pd(CH <sub>3</sub> CN) <sub>2</sub> Cl <sub>2</sub>	LiOAc	1:1	77	_
5	Pd(CH <sub>3</sub> CN) <sub>2</sub> Cl <sub>2</sub>	NaHCO <sub>3</sub>	1:0	Trace	
6	Pd(CH <sub>3</sub> CN) <sub>2</sub> Cl <sub>2</sub>	_	1:2	9	_
7	Pd <sub>2</sub> (dba) <sub>3</sub> ·CHCl <sub>3</sub> , PPh <sub>3</sub>	LiOAc	1:2	_	73
8	$Pd(PPh_3)_4$	LiOAc	1:2	—	67

Table 2. Pd(MeCN)<sub>2</sub>Cl<sub>2</sub>-catalyzed coupling reaction of 1,2-allenyl sulfoxides 1 with allyl bromide<sup>a</sup>

$ \begin{array}{c} \begin{array}{c} \begin{array}{c} \end{array} \\ \end{array} \\ \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} $ $ \begin{array}{c} \end{array} $ $ \end{array} $ $ \begin{array}{c} \end{array} $ $ \begin{array}{c} \end{array} $ $ \begin{array}{c} \end{array} $ $ \end{array} $ $ \begin{array}{c} \end{array} $ $ \begin{array}{c} \end{array} $ $ \begin{array}{c} \end{array} $ $ \end{array} $ $ \begin{array}{c} \end{array} $ $ \begin{array}{c} \end{array} $ $ \end{array} $ $ \end{array} $ $ \begin{array}{c} \end{array} $ $ \end{array} $ $ \end{array} $ $ \begin{array}{c} \end{array} $ $ \end{array} $ $ \end{array} $ $ \begin{array}{c} \end{array} $ $ \end{array} $									
Entry	1		Time (h)	Yield of <b>3</b> (%)	$E/Z^{\mathrm{b}}$				
	$R^1$	$\mathbb{R}^2$							
1	CH <sub>3</sub>	$C_2H_5$	8	73 ( <b>3b</b> )	92:8				
2	$C_2H_5$	$n-C_3H_7$	2	58 ( <b>3c</b> )	97:3				
3	$n-C_3H_7$	$n-C_4H_9$	2	78 ( <b>3d</b> )	93:7				
4	$n-C_4H_9$	$n-C_5H_{11}$	2	60 ( <b>3e</b> )	93:7				
5	(CH <sub>2</sub> ) <sub>3</sub>		9	78 ( <b>3f</b> )	93:7				
6	$(CH_{2})_{4}$		4.5	63 ( <b>3</b> g)	92:8				
7	Н	Ph	39	35 ( <b>3h</b> )	95:5				

<sup>a</sup> The reaction was carried out using 0.5 mmol of sulfoxide, 2 equiv of LiOAc, and 5 mol% of PdCl<sub>2</sub>(CH<sub>3</sub>CN)<sub>2</sub>, 1.0 mL of allyl bromide, and 1.0 mL of dry THF under N<sub>2</sub> and refluxed for the indicated time.

<sup>b</sup> E/Z Ratio referred to the 1,2-C=C bond and determined by the <sup>1</sup>H NMR analysis of the olefinic protons in the crude product.



Scheme 2.



## Scheme 3.

carbopalladation and  $\beta$ -H elimination of *E*-**5** with allyl bromide.<sup>12</sup> The stereochemistry of the 1,2-C=C bond in the product **3** may be determined by the chelation between the sulfinyl oxygen and palladium in *Z*-**5** while the configuration of 3,4-C=C bond may be determined by the possible steric repulsion between Pd and R<sup>1</sup> in **5**. The formation of products from **6**-type intermediates was not the major pathway.

In conclusion, we have observed an interesting Pd-catalyzed coupling reaction of 1,2-allenylic sulfoxides and allyl bromide, which provides an efficient entry to the synthetically useful 2-allyl-1(E),3(E)-alkadienyl sulfoxides highly stereoselectively. Due to the coordination ability of the sulfinyl group with Pd, the halopalladation is not only regioselective but also highly stereoselective. This observation may open up a new area for the regioand stereoselective intermolecular reactions of allenes. Further study in this area is being carried out in our laboratory.

## Acknowledgements

Financial supports from the Major State Basic Research Development Program (Grant No G2000077500), National Natural Science Foundation of China, and Shanghai Municipal Committee of Science and Technology are greatly appreciated.

## **References and notes**

 (a) Schultz, R. G. *Tetrahedron Lett.* **1964**, *5*, 301–304; (b) Schultz, R. G. *Tetrahedron* **1964**, *20*, 2809–2813.

- Lupin, M. S.; Shaw, B. L. Tetrahedron Lett. 1964, 5, 883– 885.
- (a) Hegedus, L. S.; Kambe, N.; Ishii, Y.; Mori, A. J. Org. Chem. 1985, 50, 2240–2243; (b) Hegedus, L. S.; Kambe, N.; Tamura, R.; Woodgate, P. D. Organometallics 1983, 2, 1658–1661.
- For reviews on carbopalladation of allenes, see: Zimmer, R.; Danesh, C. U.; Nadanan, E.; Hhan, F. A. *Chem. Rev.* 2000, 111, 3067–3126; Ma, S. Acc. Chem. Res. 2003, 36, 701–712.
- 5. Tsuji, J.; Susuki, T. Tetrahedron Lett. 1965, 3027-3031.
- (a) Alper, H.; Hartstock, F. W.; Despeyroux, B. J. Chem. Soc., Chem. Commun. 1984, 905–906; (b) Fergusson, S. B.; Alper, H. J. Mol. Catal. 1986, 34, 381–384.
- For some of the most recent results from this group, see:
   (a) Ma, S.; Zhao, S. J. Am. Chem. Soc. 2001, 123, 5578– 5579; (b) Ma, S.; Shi, Z. Chem. Commun. 2002, 540–541;

(c) Ma, S.; Gao, W. Synlett. 2002, 65–68; (d) Ma, S.; Duan, D.; Wang, Y. J. Comb. Chem. 2002, 4, 239–247; (e) Ma, S.; Yu, Z. Angew. Chem., Int. Ed. 2002, 41, 1775–1778; (f) Ma, S.; Jiao, N. Angew. Chem., Int. Ed. 2002, 41, 4737–4740; (g) Ma, S.; Yu, Z. Angew. Chem., Int. Ed. 2003, 42, 1955–1957.

- 8. For a most recent review on the chemistry of 1,2-allenylic sulfoxides, see: Wei, Q.; Ma, S. *Chin. J. Org. Chem.* **2002**, 22, 254–261.
- 9. Ma, S.; Wei, Q.; Wang, H. Org. Lett. 2000, 2, 3893-3895.
- 10. Ma, S.; Ren, H.; Wei, Q. J. Am. Chem. Soc. 2003, 125, 4817-4830.
- Kaneda, K.; Uchiyama, T.; Fujiwara, Y.; Imanaka, T.; Teranishi, T. J. Org. Chem. **1979**, 44, 55–63; Ma, S.; Lu, X. J. Chem. Soc., Chem. Commun. **1990**, 733–734; Ma, S.; Gao, W. J. Org. Chem. **2002**, 67, 6104–6112.
- 12. Heck, R. F. Org. React. 1982, 27, 345-390.